

**COATING COMPOSITIONS PROVIDING IMPROVED MAR AND
SCRATCH RESISTANCE AND METHODS OF USING THE SAME**

5

CROSS REFERENCE TO RELATED APPLICATIONS

This application claims priority to provisional application 60/254,143,
filed on December 8, 2000.

10

FIELD OF THE INVENTION

The present invention relates to coating compositions that provide
improved mar and/or scratch resistance and to methods for using the same.
More specifically, the improved resistance is achieved by adding particles to a
film forming resin.

15

BACKGROUND OF THE INVENTION

"Color-plus-clear" coating systems involving the application of a colored
or pigmented basecoat to a substrate followed by application of a transparent
or clearcoat over the basecoat have become increasingly popular as original
20 finishes for a number of consumer products including, for example, cars and
floor coverings such as ceramic tiles and wood flooring. The color-plus-clear
coating systems have outstanding appearance properties such as gloss and
distinctness of image, due in large part to the clear coat.

"One coat" systems comprising a one coat color layer are applied
25 themselves as the topcoat. One coat systems are frequently used for
household appliances, lawn and garden equipment, interior fixtures, and the
like.

In recent years, powder coatings have become increasingly popular
because these coatings are inherently low in volatile organic content (VOC),
30 which significantly reduces air emissions during the application and curing
processes. Liquid coatings are still used in many systems, however,
particularly those wherein solvent emissions are permitted. For example, the
coating of elastomeric automotive parts is often done by spraying liquid
compositions. Many of these compositions are formulated to be flexible so

the coating can bend or flex with the substrate without cracking. Because these coatings can result in films that are softer, they may be more susceptible to marring and scratching.

- Topcoat film-forming compositions, such as the protective and/or decorative one coats for household appliances and the transparent clearcoat in color-plus-clear coating systems for automotive applications, are subject to defects that occur during the assembly process and damage from both the environment and normal use of the end product. Paint defects that occur during assembly include the paint layer being too thick or too thin, "fish eyes" or craters, and under-cured or over-cured paint; these defects can affect the color, brittleness, solvent resistance and mar and scratch performance of the coating. Damaging environmental factors include acidic precipitation, exposure to ultraviolet radiation from sunlight, high relative humidity and high temperatures; these factors can also result in compromised performance.
- Normal use of consumer products will often lead to marring, scratching and/or chipping of the surface due to contact with hard objects, contact with brushes and/or abrasive cleansers during normal cleaning processes, and the like.

Thus, there is a need in the coatings art for topcoats having good scratch and mar resistance, including those in which flexibility would also be desired.

SUMMARY OF THE INVENTION

- The present invention is directed to coating compositions general comprising a film-forming resin in which is dispersed a plurality of particles.
- The particles can be organic or inorganic particles, or mixtures thereof. The particles typically have an average particle size ranging from 0.1 to 15 microns. Methods for using these compositions are also within the scope of the invention, as are articles coated according to these methods.

It has been surprisingly discovered that the incorporation of particles into a film-forming resin results in coatings having enhanced mar and/or

scratch resistance as compared with the same coatings lacking these particles. According to the present invention, coatings can be formulated with these improved mar and/or scratch characteristics without affecting the appearance or other mechanical properties of the coatings.

5 "Mar" and "scratch" refer herein to physical deformations resulting from mechanical or chemical abrasion. "Mar resistance" is a measure of a material's ability to resist appearance degradation caused by small scale mechanical stress. "Scratch resistance" is the ability of a material to resist more severe damage that can lead to visible, deeper or wider trenches. Thus, 10 scratches are generally regarded as being more severe than what is referred to in the art as mar, and the two are regarded in the art as different. As noted above, mar and scratch can result from manufacturing and environmental factors as well as through normal use. Although mar and scratch are in many respects differing degrees of the same thing, a coating that improves mar 15 resistance may not be effective in improving scratch resistance, and vice versa. It will be appreciated, therefore, that combinations of particles can be employed to give the final coating its desired characteristics. For example, one particle that offers particularly good mar resistance can be coupled with one that offers particularly good scratch resistance.

20

DESCRIPTION OF THE INVENTION

The present invention is directed to a coating comprising a film-forming resin and a plurality of particles dispersed in the resin.

Any resin that forms a film can be used according to the present 25 methods, absent compatibility problems. For example, resins suitable for both powder and liquid coating compositions can be employed.

A particularly suitable resin for use in the present powder compositions is one formed from the reaction of a polymer having at least one type of reactive functional group and a curing agent having functional groups reactive 30 with the functional group of the polymer. The polymers can be, for example,

acrylic, polyester, polyether or polyurethane, and can contain functional groups such as hydroxyl, carboxylic acid, carbamate, isocyanate, epoxy, amide and carboxylate functional groups.

5 The use in powder coatings of acrylic, polyester, polyether and polyurethane polymers having hydroxyl functionality is known in the art. Monomers for the synthesis of such polymers are typically chosen so that the resulting polymers have a glass transition temperature (" T_g ") greater than 50°C. Examples of such polymers are described in U.S. Patent No. 5,646,228 at column 5, line 1 to column 8, line 7, incorporated by reference herein.

10 Acrylic polymers and polyester polymers having carboxylic acid functionality are also suitable for powder coatings. Monomers for the synthesis of acrylic polymers having carboxylic acid functionality are typically chosen such that the resulting acrylic polymer has a T_g greater than 40°C, and for the synthesis of the polyester polymers having carboxylic acid functionality
15 such that the resulting polyester polymer has a T_g greater than 50°C. Examples of carboxylic acid group-containing acrylic polymers are described in U.S. Patent No. 5,214,101 at column 2, line 59 to column 3, line 23, incorporated by reference herein. Examples of carboxylic acid group-containing polyester polymers are described in U.S. Patent No. 4,801,680 at
20 column 5, lines 38 to 65, incorporated by reference herein.

The carboxylic acid group-containing acrylic polymers can further contain a second carboxylic acid group-containing material selected from the class of C_4 to C_{20} aliphatic dicarboxylic acids, polymeric polyanhydrides, low molecular weight polyesters having an acid equivalent weight from about 150
25 to about 750, and mixtures thereof. This material is crystalline and is preferably a low molecular weight crystalline carboxylic acid group-containing polyester.

Also useful in the present powder coating compositions are acrylic, polyester and polyurethane polymers containing carbamate functional groups.
30 Examples are described in WO Publication No. 94/10213, incorporated by

reference herein. Monomers for the synthesis of such polymers are typically chosen so that the resulting polymer has a high T_g , that is, a T_g greater than 40°C . The T_g of the polymers described above can be determined by differential scanning calorimetry (DSC).

- 5 Suitable curing agents generally include blocked isocyanates, polyepoxides, polyacids, polyols, anhydrides, polyamines, aminoplasts and phenoplasts. The appropriate curing agent can be selected by one skilled in the art depending on the polymer used. For example, blocked isocyanates are suitable curing agents for hydroxy and primary and/or secondary amino
- 10 group-containing materials. Examples of blocked isocyanates are those described in U.S. Patent No. 4,988,793, column 3, lines 1 to 36, incorporated by reference herein. Polyepoxides suitable for use as curing agents for COOH functional group-containing materials are described in U.S. Patent No. 4,681,811 at column 5, lines 33 to 58, incorporated by reference herein.
- 15 Polyacids as curing agents for epoxy functional group-containing materials are described in U.S. Patent No. 4,681,811 at column 6, line 45 to column 9, line 54, incorporated by reference herein. Polyols, materials having an average of two or more hydroxyl groups per molecule, can be used as curing agents for NCO functional group-containing materials and anhydrides, and
- 20 are well known in the art. Polyols for use in the present invention are typically selected such that the resultant material has a T_g greater than about 50°C .

 Anhydrides as curing agents for epoxy functional group-containing materials include, for example, trimellitic anhydride, benzophenone tetracarboxylic dianhydride, pyromellitic dianhydride, tetrahydrophthalic

25 anhydride, and the like as described in U.S. Patent No. 5,472,649 at column 4, lines 49 to 52, incorporated by reference herein. Aminoplasts as curing agents for hydroxy, COOH and carbamate functional group-containing materials are well known in the art. Examples of such curing agents include aldehyde condensates of glycoluril, which give high melting crystalline

30 products useful in powder coatings. While the aldehyde used is typically

formaldehyde, other aldehydes such as acetaldehyde, crotonaldehyde, and benzaldehyde can be used.

The film-forming resin described above is generally present in the present powder coating compositions in an amount greater than about 50
5 weight percent, such as greater than about 60 weight percent, and less than 90 weight percent, with weight percent being based on the total weight of the composition. For example, the weight percent of resin can be between 60 and 70 weight percent. When a curing agent is used, it is generally present in an amount of between about 10 and 40 weight percent; this weight percent is
10 also based on the total weight of the coating composition.

The present compositions can be formed from film-forming resins that are liquid, that is, water-borne or solvent-borne systems. Organic solvents in which the present coatings may be dispersed include, for example, alcohols, ketones, aromatic hydrocarbons, glycol ethers, esters or mixtures thereof.
15 Examples of polymers useful in forming the resin in the liquid coatings of the present invention include hydroxyl or carboxylic acid-containing acrylic copolymers, hydroxyl or carboxylic acid-containing polyester polymers, oligomers and isocyanate or hydroxyl-containing polyurethane polymers, and amine or isocyanate-containing polyureas. These polymers are further
20 described in U.S. Patent No. 5,939,491, column 7, line 7 to column 8, line 2; this patent, as well as the patents referenced therein, are incorporated by reference herein. Curing agents for these resins are also described in the '491 patent at column 6, line 6 to line 62. In solvent-based compositions, the solvent is generally present in amounts ranging from 5 to 80 weight percent
25 based on total weight of the composition, such as 30 to 50 percent.

Any combination of organic or inorganic particles can be added to the resin according to the present invention. Examples of organic particles include diamond particles, such as diamond dust particles, and particles formed from carbide materials; examples of carbide particles include but are
30 not limited to titanium carbide, silicon carbide and boron carbide. Inorganic

particles include but are not limited to silica; alumina; alumina silicate; silica alumina; alkali aluminosilicate; borosilicate glass; nitrides including boron nitride and silicon nitride; oxides including titanium dioxide and zinc oxide; quartz; nepheline syenite; zircon such as in the form of zirconium oxide; buddeluyite; and eudialyte. Mixtures of any of the above particles can be used, including different combinations of organic particles, inorganic particles, or both. The silica can be in any suitable form, such as crystalline, amorphous, or precipitated; crystalline silica is particularly suitable for one-coat applications. The alumina can be used in any of its forms, such as alpha, beta, gamma, delta, theta, tabular alumina, and the like and can be fused or calcined, and if calcined, ground or unground. Alpha alumina having a crystalline structure is particularly suitable for clear coats used in the automotive industry.

The particles listed above are widely commercially available. For example, crystalline silica is available from Reade Advanced Materials; amorphous and precipitated silica from PPG Industries, Inc.; ZEEOSPHERES, silica alumina ceramic alloy particles, from 3M Corporation; silica alumina, such as G200, G-400, G-600, from 3M Corporation; alkali alumina silicate, such as W-210, W-410, and W-610, from 3M Corporation; borosilicate glass, sold as SUNSPHERES, from MoSci Corporation; and quartz and nepheline syenite from Unimin, Inc. Tabular alumina is available from Micro Abrasives Corporation as WCA3, WCA3S, and WCA3TO, and from Alcoa as T64-20. Zircon, buddeluyite and eudialyte are commercially available from Aran Isles Corporation, and boron nitride is available from Carborundum Inc. as SHP-605 and HPP-325. It will be appreciated that many commercially available products are actually composites or alloys of one or more materials; such particles are equally within the scope of the present invention.

In some embodiments, it might be desirable to heat treat the particles before incorporating them into the present compositions. Heat treating can be

accomplished, for example, by heating the particles at a temperature of between about 350°C and 2000°C, such as 600°C and 1000°C for a time period of two to three hours.

5 The particles used in the present invention have an average particle size ranging from about 0.1 to 15 microns, such as from 1 to 12 microns, 1 to 10 microns, or 3 to 6 microns. Any of the particles listed above can be used in any size within these ranges according to the present invention. In one embodiment, the particle size is less than 10 microns. In one embodiment, the particles are silicon carbide, calcined alumina or tabular alumina having a median particle size range of less than 6 microns, such as less 5.5 microns or even less than 3 microns. In one embodiment the particles are unground calcined alumina having a median crystallite size of less than 5.5 microns, such as about 2 microns. "Average particle size" refers to the size of about 50 percent or more of the particles in a sample. "Median particle size" refers to 15 the particle size at which half of the distribution is larger and half is smaller; "median crystallite size" is similarly defined, but using the crystallite size rather than the particle size.

Particle size can be determined according to any method known in the art, such as by a conventional particle size analyzer. For example, where the 20 average particle size is greater than 1 micron laser scattering techniques can be employed, and for average particle sizes smaller than 1 micron, transmissional electron microscopy ("TEM") can be used.

The shape or morphology of the particles can vary depending on the type of particle or particles selected. For example, generally spherical 25 particles, such as crystalline materials, solid beads, microbeads, or hollow spheres, can be used, as can particles that are platy, cubic or acicular (that is, elongated or fibrous). The particles can also have a random or nonuniform morphology. In addition, the particles can have an internal structure that is hollow, porous or void free, or any combination, such as a hollow center with 30 porous or solid walls. It will be appreciated that different particle shapes may

be more suitable for one application over another. For example, when used with automotive clearcoats, particles having a platy morphology may have better mar resistance than those having spherical or other nonspherical forms. Particle shape may be irrelevant, however, for other applications. It will be appreciated that combinations of particles having different morphologies can be used to give the desired characteristics to the final coating.

The particles should have a hardness sufficient to impart greater protection from mar and/or scratch than would be achieved in a coating made from the same resin but lacking the particles. For example, the particles can have a hardness value greater than the hardness value of materials that can scratch or mar a cured coating, such as dirt, sand, rocks, glass, abrasive cleaners, car wash brushes, and the like. The hardness value of the particles and materials that can scratch or mar a coating can be determined by any conventional hardness measurement method, but is typically determined according to the Mohs hardness scale. The Mohs scale is an empirical scale of the hardness of minerals or mineral-like materials, and indicates the relative scratch resistance of the surface of a material. The original Mohs scale consisted of the values ranging from 1 to 10, with talc having a value of 1 and diamond having a value of 10. The scale has recently been expanded from a maximum value of 10 to a maximum value of 15 to accommodate the addition of some synthetic materials. All of the Mohs hardness values discussed herein, however, are based upon the original 1 to 10 scale.

The Mohs hardness values of several particles within the scope of the invention are given in Table A below.

Table A

PARTICLE MATERIAL	MOHS HARDNESS
Silicon nitride	9+
Zinc oxide	4.5
Crystalline silica	6.5 - 7.0
Titanium carbide	9.0
α -alumina	9.0
γ -alumina	8.0
Borosilicate glass	4.5 - 6.5
Diamond	10.0
Boron carbide	9.7

Typically, the particles used according to the present invention will have a Mohs hardness of about 4.5 or greater, such as about 5 or greater.

- 5 For automobile clearcoats, particles having a Mohs hardness of 9 or 10 is often the most suitable. In one embodiment, the Mohs hardness of the particles is between 4.5 and 8, such as between 4.5 and 7.5, or 4.5 and 7.

- It will be appreciated that many particles, particularly the inorganic particles, according to the present invention have a hardness at their surface that can be different from the hardness of the internal portions of the particle. The hardness of the surface is typically the hardness relevant to the present invention.
- 10

- As noted above, the particles or combination of particles used in the present invention should generally have a hardness sufficient to impart improved protection from mar and/or scratch as compared to no particle being present. Accordingly, the present compositions, when cured, will have greater mar and/or scratch resistance than their particle-lacking counterparts. Gloss retention percentages following mar and/or scratch testing ranging from 20 percent up to near 100 percent are achieved, such as 20 percent or greater retention, 50 percent or greater retention, or 70 percent or greater retention. To determine whether improved mar and scratch resistance is obtained with a
- 15
- 20

particular particle or combination of particles, two coating compositions can be formulated, with the only difference being that one contains the present particles and one does not. The coatings can be tested for mar and/or scratch resistance (i.e. "mar and/or scratch testing") by any means standardly known in the art, such as those described in the Example section below. The results for the particle-containing and nonparticle-containing compositions can be compared to determine whether improved resistance is obtained when the selected particles are added. Even a small improvement in any of these tests constitutes an improvement according to the invention. It will be appreciated that mar and scratch resistance, and methods for testing the same, are distinct from "wear-through", weight loss, or bulk-film properties tested, for example, using a Taber abraser, and that such tests are typically relevant to products other than those of the present invention.

The particles are typically present in the curable coating composition of the present invention in an amount ranging from 0.1 to 20.0 weight percent, such as from 0.1 to 10 weight percent, or from 0.1 to 8 weight percent, with weight percent based on total weight of the coating composition. In one embodiment, the particles are present in a concentration of greater than 5 weight percent, such as greater than 5 up to 20 weight percent. While amounts of 20 weight percent or less are typically suitable, amounts even greater than 20 weight percent can also be used. It will be appreciated that improvement in mar and scratch resistance will increase as the concentration of particles increases. The tests described in the Example section below can be used by those skilled in the art to determine what weight percent or "load" of particles will give the desired level of protection. The particles will be fairly evenly dispersed in the cured coating, that is, there will not typically be an increased concentration of particles in one portion of the cured coating as compared with another.

Both the size of the particles used as well as the particle load can affect not only the level of mar and/or scratch resistance but also the

appearance of the cured coating. Thus, particle size and load should be optimized by the user based on the particular application, taking into account, for example, the level of acceptable haze, the level of mar and/or scratch resistance, the thickness of the coating and the like. Where appearance is particularly relevant, such as in an automotive clear coat, a relatively low load and particle size can be used. A load of less than 5 weight percent, even less than 1 weight percent, and a particle size between about 3 to 6 microns is particularly suitable. For industrial one-coat systems where haze is not as relevant, or where other pigments are present, loadings of up to about 10 percent or even higher can be used, as can particle sizes of 10 microns or even larger. One skilled in the art can optimize particle size and load to achieve the desired level of mar and/or scratch resistance without compromising the appearance or other mechanical properties of the cured coatings. Mixtures of particles having different sizes may be particularly suitable for a given application.

Haze can also be minimized to at least some degree by selecting resins and particles that have a similar refractive index ("RI"), that is the difference between the resin RI and the particle RI (" Δ RI") is minimized. In some applications, such as for clear coats, the Δ RI can be less than one, or even less than 0.1. Using a combination of particles having different RI's can also help to reduce haze. Minimizing Δ RI is particularly relevant when the particles are larger in size (i.e. greater than about 6 microns) and/or the particle load is greater than about 8 weight percent. When the RI of the particle is close to the RI of the resin, the particles may comprise greater than 20 weight percent of the present compositions.

In another embodiment of the present invention, in addition to the particles described above, nanoparticles are also incorporated into the present compositions. "Nanoparticles" is used herein to refer to particles having an average particle size from 0.8 to less than 500 nanometers, such as between 10 and 100 nanometers. Such nanoparticles can include both

organic and inorganic particulate materials, such as those formed from polymeric and nonpolymeric organic and inorganic materials, composite materials, and mixtures thereof. As used herein, the term "polymeric inorganic material" means a polymeric material having a backbone repeat unit
5 based on an element or elements other than carbon, for example silicon; "polymeric organic materials" means synthetic polymeric materials, semisynthetic polymeric materials and natural polymeric materials, all of which have a backbone repeat unit based on carbon. "Composite material" refers to a combination of two or more different materials that have been
10 combined. The nanoparticles formed from composite materials can have a hardness at their surface that is different from the hardness of the internal portions of the particle. The surface of the nanoparticles can be modified such as by chemically or physically changing its surface characteristics using techniques known in the art. For example, the nanoparticles can be
15 dispersed in siloxane, such as one to which an acid functional group has been added. In addition, a nanoparticle formed from one material can be coated, clad or encapsulated with a different material or different form of the same material to yield a particle having the desired surface characteristics.

The nanoparticles suitable for use in the compositions of the invention
20 can be formed from ceramic materials, metallic materials, or mixtures thereof or can comprise, for example, a core of essentially a single inorganic oxide such as silica in colloidal, fumed, or amorphous form, alumina or colloidal alumina, titanium dioxide, cesium oxide, yttrium oxide, colloidal yttrium, zirconia such as colloidal or amorphous zirconia or mixtures thereof, or an
25 inorganic oxide of one type upon which is deposited an organic oxide of another type. Materials useful in forming the present nanoparticles include graphite, metals, oxides, carbides, nitrides, borides, sulfides, silicates, carbonates, sulfates and hydroxides.

As discussed above, in many applications it will be desired that the use
30 of the present particles and, when employed, the nanoparticles should not

significantly interfere with the optical properties of the cured coating composition. Haze can be determined using a BYK/Haze Gloss instrument. The haze of a cured coating both with and without the present particles ("Δ haze value") of less than about 10 or even lower is typically desired for most applications. A Δ haze value of 5 or less is typically desired when using the present compositions as a transparent topcoat.

The powder coating compositions of the present invention may optionally contain additives such as waxes for flow and wetting, flow control agents, such as poly(2-ethylhexyl)acrylate, degassing additives such as benzoin and MicroWax C, adjuvant resin to modify and optimize coating properties, antioxidants, ultraviolet (UV) light absorbers and catalysts. Examples of useful antioxidants and UV light absorbers include those available commercially from Ciba-Geigy under the trademarks IRGANOX® and TINUVIN®. These optional additives, when used, are typically present in amounts up to 20 percent by weight, based on total weight of the coating.

The liquid compositions of the present invention can similarly contain optimal additives such as plasticizers, antioxidants, light stabilizers, UV absorbers, thixotropic agents, anti-gassing agents, organic cosolvents, biocides, surfactants, flow control additives and catalysts. Any such additives known in the art can be used, absent compatibility problems.

The particles of the present invention can be added at any time during the formulation of the powder or liquid coating. For example, curable powder coating compositions of the present invention can be prepared by first dry blending the film-forming resin, the plurality of particles, and any of the additives described above, in a blender, such as a Henschel blade blender. The blender is operated for a period of time sufficient to result in a homogenous dry blend of the materials. The blend is then melt blended in an extruder, such as a twin screw co-rotating extruder, operated within a temperature range sufficient to melt but not gel the components. The melt blended curable powder coating composition is typically milled to an average

particle size of from, for example, 15 to 80 microns. Other methods known in the art can also be used.

Alternatively, the present powder compositions can be prepared by blending and extruding the ingredients as described above, but without the present particles. The particles can be added as a post-additive to the formulation, such as through a second extrusion process or by simply mixing the particles into the blended composition, such as by shaking them together in a closed container or using a Henschel mixer. While compositions comprising post-added particles have been surprisingly found to give better mar and/or scratch resistance, ease of use, processibility and appearance are often better when the particles are incorporated into the formulation with the other dry ingredients. The manner of formulating the present compositions can therefore be determined by one skilled in the art depending on the application and desired parameters of the user.

The coating compositions of the invention can be applied to a variety of substrates, for example automotive substrates such as fenders, hoods, doors and bumpers, and industrial substrates such as household appliances, including washer and dryer panels and lids, refrigerator doors and side panels, lighting fixtures and metal office furniture. Such automotive and industrial substrates can be metallic, for example, aluminum and steel substrates, and non-metallic, for example, thermoplastic or thermoset (i.e. "polymeric") substrates.

The powder coating compositions are most often applied by spraying, and in the case of a metal substrate, by electrostatic spraying, or by the use of a fluidized bed. The powder coating can be applied in a single sweep or in several passes to provide a film having a thickness after cure of from about 1 to 10 mils (25 to 250 micrometers), usually about 2 to 4 mils (50 to 100 micrometers). Other standard methods for coating application can be employed such as brushing, dipping or flowing.

The liquid compositions of the invention can also be applied by any conventional method such as brushing, dipping, flow coating, roll coating, conventional and electrostatic spraying. Spray techniques are most often used. Typically, film thickness for liquid coatings can range between 0.1 and 5 mils, such as between 0.1 and 1 mil, or about 0.4 mils.

Generally, after application of the coating composition, the coated substrate is baked at a temperature sufficient to cure the coating. Metallic substrates with powder coatings are typically cured at a temperature ranging from 250°F to 500°F (121.1°C to 260.0°C) for 1 to 60 minutes, or from 300°F to 400°F (148.9°C to 204.4°C) for 15 to 30 minutes.

Several liquid formulations can be cured at ambient temperature, such as those using a polyisocyanate or polyanhydride curing agent, or they can be cured at elevated temperatures to hasten the cure. An example would be forced air curing in a down draft booth at about 40°C to 60°C, which is common in the automotive refinish industry. The ambient temperature curable compositions are usually prepared as a two (2) package system in which the curing agent is kept separate from the polysiloxane containing the reactive functional group. The packages are combined shortly before application.

The thermally curable liquid compositions such as those using blocked isocyanate, aminoplast, phenoplast, polyepoxide or polyacid curing agent can be prepared as a one-package system. These compositions are cured at elevated temperatures, typically for 1 to 30 minutes at about 250°F to about 450°F (121°C to 232°C) with temperature primarily dependent upon the type of substrate used. Dwell time (i.e., time that the coated substrate is exposed to elevated temperature for curing) is dependent upon the cure temperatures used as well as wet film thickness of the applied coating composition. For example, coated automotive elastomeric parts require a long dwell time at a lower cure temperature (e.g., 30 minutes 250°F (121°C), while coated

aluminum beverage containers require a very short dwell time at a very high cure temperature (e.g., 1 minute 375°F (191°C)).

The coating compositions of the invention are particularly useful as primers and as color and/or clear coats in color-clear composite coatings.

- 5 The compositions of the invention in the pigmented form can be applied directly to a substrate to form a color coat. The color coat may be in the form of a primer for subsequent application of a top coat or may be a colored top coat. Alternately, the coating composition of the invention can be unpigmented, in the form of a clearcoat for application over a color coat
- 10 (either a primer coat or a colored topcoat). When used as a primer coating, thicknesses of 0.4 to 4.0 mils are typical. When used as a color topcoat, coating thicknesses of about 0.5 to 4.0 mils are usual, and when used as a clearcoat, coating thicknesses of about 1.5 to 4.0 mils are generally used.

- Accordingly, the present invention is further directed to a substrate
- 15 coated with one or more of the present compositions. The substrates and compositions, and manner of applying the same, are as described above.

- The present invention is further directed to a multi-layer composite coating composition comprising a base coat deposited from a film-forming composition and a topcoat applied over at least a portion of the base coat,
- 20 where the topcoat is deposited from any of the coating compositions of the present invention. The base coat might have a cured film thickness between about 0.5 to 4 mils (12.5 to 100 micrometers) while the topcoat cured film thickness can be up to 10 mils (250 micrometers). The base coat can be cured before application of the topcoat, or the two coats can be cured
- 25 together. In one example, the base coat can be deposited from a pigmented film-forming composition, while the topcoat formed from the present compositions is substantially transparent. This is the color-plus-clear system discussed above, frequently used in automotive applications.

- In yet another embodiment, the present invention is directed to a
- 30 method for improving the mar and/or scratch resistance of a coated substrate

comprising applying the present compositions to at least a portion of the substrate. Application can be by any means known in the art to the thicknesses described above.

The coatings formed according to the present invention have
5 outstanding appearance properties and scratch and mar resistance properties as compared to no particles being present. It has been surprisingly discovered that the compositions of the present invention result in coatings having exceptional resistance to UV degradation. Accordingly, the invention is further directed to a cured coating having particles dispersed throughout,
10 such as a powder coating, having less than 10 percent, such as less than 5 percent or even less than 4 percent, reduction in gloss after 500, 1000, and 1500 hours of QUV exposure. As shown in the Examples below, the coatings of the present invention can even have improved resistance following QUV exposure. "QUV exposure" refers to any type of QUV exposure, such as
15 testing done pursuant to ASTM D-4587.

As used herein, unless otherwise expressly specified all numbers such as those expressing values, ranges, amounts or percentages may be read as if prefaced by the word "about", even if the term does not expressly appear. Also, any numerical range recited herein is intended to include all sub-ranges
20 subsumed therein. As used herein, the term "polymer" refers to oligomers and both homopolymers and copolymers.

EXAMPLES

For all of the Examples, unless otherwise noted, 20° gloss was
25 measured with a handheld 20° NOVO-GLOSS 20 statistical glossmeter, available from Gardener Instrument Company, Inc.

BON AMI Mar Resistance ("BON AMI") was performed using an Atlas AATCC Mar Tester Model CM-5, available from Atlas Electrical Devices Co. of Chicago, Illinois. Using a felt cloth clamped to the acrylic finger on the arm
30 of the instrument, a set of 10 double rubs (unless indicated otherwise) was

run on each panel, which was coated with BON AMI cleanser. The panel was then washed with cool tap water and dried. In the tables below, mar resistance is expressed as a percentage of the 20° gloss that was retained after the surface was marred by the mar tester. Mar resistance was

5 measured as: $\text{Mar Resistance} = (\text{Marred Gloss} \div \text{Original Gloss}) \times 100$.

1, 2, and 9μ 3M Abrasive Paper Scratch Resistance ("1, 2 or 9μ Paper") also was performed using the Atlas Tester. A 2" x 2" piece of the 3M Abrasive Paper backed with the felt cloth was clamped to the acrylic finger on the arm of the instrument, and a set of 10 double rubs (unless indicated

10 otherwise) was run on each panel. The panel was then washed with cool tap water and dried. In the tables below, scratch resistance is expressed as the percentage of the 20° gloss that was retained after the surface was scratched by the scratch tester. Scratch resistance was measured as: $\text{Scratch Resistance} = (\text{Scratched Gloss} \div \text{Original Gloss}) \times 100$.

15 BYK Gardner haze was measured using the BYK/Haze Gloss Instrument following manufacturer's instructions.

Steel wool tests were also performed using the Atlas Tester ("steel wool") in the same manner as the scratch tests only using a 2" x 2" piece of the 0000# grade steel wool sheet backed with the felt cloth.

20 Steel wool tests were also performed using a light hammer (571 grams "light hammer") or heavy hammer (1381 grams "heavy hammer") wrapped with 0000# grade steel wool. In some cases, the heavy hammer had a 1382 gram weight mounted on top. These tests were otherwise performed as described above for the scratch tests. Values repaired in the tables below for
25 the steel wool tests are percent gloss retention.

The following examples are intended to illustrate the invention, and should not be construed as limiting the invention in any way.

Example 1

Epoxy-acid powder clear coat compositions identified as Samples 1 through 7 in Table I were prepared using the components and amounts (parts by weight) shown, and processed in the following manner. The components
5 were blended in a Henschel Blender for 60 to 90 seconds. The mixtures were then extruded through a Werner & Pfleider co-rotating twin screw extruder at a 450 RPM screw speed and an extrudate temperature of 100°C to 125°C. The extruded material was then ground to a particle size of 17 to 27 microns using an ACM Grinder (Air Classifying Mill from Micron Powder Systems,
10 Summit, New Jersey). Cold rolled steel test panels were coated with PPG Black Electrocoat primer ED5051, fully cured, and were obtained from ACT Laboratories. The finished powders were electrostatically sprayed onto test panels and evaluated for coatings properties as discussed below.

100440001

Table 1

Description	Sample 1 Control Formula (no particles)	Sample 2 0.1% Diamond Dust MBM 4-8	Sample 3 0.3% W610 Zeospheres	Sample 4 0.3% WCA-3 alumina platelets	Sample 5 0.3% WCA-3 + 1.0% Sunospheres 05	Sample 6 1.0% Sunospheres 05	Sample 7 0.3% W210 Zeospheres
GMA Functional Acrylic ¹	69.05	68.98	68.83	68.83	68.08	68.30	68.83
DDDA ²	22.68	22.65	22.60	22.60	22.35	22.43	22.60
Benzoin	0.20	0.20	0.20	0.20	0.20	0.20	0.20
Wax C Micropowder ³	0.60	0.60	0.60	0.60	0.60	0.60	0.60
Tinuvin 144 ⁴	2.00	2.00	2.00	2.00	2.00	2.00	2.00
CGL-1545 ⁵	2.00	2.00	2.00	2.00	2.00	2.00	2.00
HCA-1 ⁶	2.00	2.00	2.00	2.00	2.00	2.00	2.00
ARMEEN M2C ⁷	0.37	0.37	0.37	0.37	0.37	0.37	0.37
Acrylic Flow Additive ⁸	1.10	1.10	1.10	1.10	1.10	1.10	1.10

¹ 50% Glycidal Methacrylate/10% Butyl Methacrylate/5% Styrene/35% Methyl Methacrylate.

² Dodecanedioic acid.

³ Wax C Micro Powder, a fatty acid amide (ethylene bis-stearoylamide), commercially available from Hoechst-Celanese.

⁴ TINUVIN 144 (2-tert-butyl-2-(4-hydroxy-3,5-di-tert-butylbenzyl)[bis(methyl-2,2,6,6-tetramethyl-4-piperidyl)]dipropionate), an ultraviolet light stabilizer available from Ciba-Geigy Corp.

⁵ CGL-1545 (2-[4((2-Hydroxy-3-(2-ethylhexyloxy)propyl)-oxy]-2-hydroxyphenyl)-4,6-bis(2,4-dimethylphenyl)-1,3,5-triazine), an ultraviolet light stabilizer available from Ciba-Geigy Corp.

⁶ HCA-1, an anti-yellowing agent (antioxidant) commercially available from Sanko Chemical Corp.

⁷ Methyl dicocaoamine available from Akzo-Nobel Corp., used as a catalyst.

⁸ Acrylic Flow Agent prepared by solution polymerization in xylene and toluene of the following monomers: 81.2% 2-ethyl hexyl acrylate, 11.8% hydroxyl ethyl acrylate, and 7% N,N-dimethylaminoethyl methacrylate. The polymerization was at reflux temperature in the presence of VAZO 67 (2,2'-Azobis-(2-methylbutyronitrile)). The acrylic flow agent was vacuum stripped to 100% solids.

Table 1 (Cont'd)

Description	Sample 1 Control Formula (no particles)	Sample 2 0.1% Diamond Dust MBM 4-8	Sample 3 0.3% W610 Zeospheres	Sample 4 0.3% WCA-3 alumina platelets	Sample 5 0.3% WCA-3 + 1.0% Sunospheres 05	Sample 6 1.0% Sunospheres 05	Sample 7 0.3% W210 Zeospheres
Diamond Dust MBM 4-8 ⁹		0.10					
Zeospheres W610 ¹⁰			0.30				
WCA-3 alumina platelets ¹¹				0.30			
Zeospheres W210 ¹²					1.00	1.00	0.30
Sunospheres 05 ¹³					100.00	100.00	
Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00

⁹ Diamond Dust MBM 4-8 is commercially available from GE Superabrasives, average particle size 6 microns.

¹⁰ Zeospheres W-610, solid spheres of a silica-alumina ceramic alloy, are commercially available from 3M Corporation, average particle size 10 microns.

¹¹ WCA-3 alumina platelets are commercially available from Micro Abrasives Corporation, average particle size 3 microns.

¹² Zeospheres W210, solid spheres of a silica-alumina ceramic alloy, are commercially available from 3M Corporation, average particle size 3 microns.

¹³ Sunospheres 05, fused quartz/borosilicate glass microspheres, are commercially available from MO-SCI Corporation, average particle size 0.6 microns.

The powder coatings of Samples 1-7 were applied at 2.3 to 2.8 mils (58 to 71 microns) and cured for 30 minutes at 293°F (145°C). The panels were then subjected to the mar and scratch tests indicated in the tables below. A number of control panels were prepared. Generally, Samples 2-7, which represent compositions according to the present invention, performed better in all tests as compared to Sample 1, which did not include the present particles.

Table 2

Mar/Scratch Resistance Test	Sample 1 Control	Sample 2 0.1% Diamond Dust	Sample 6 1.0% Sunspheres	Sample 7 0.3% Zeeospheres W-210
BON AMI	57	61	79	83
1μ paper	63	90	67	86
2μ paper	52	91	63	72
9μ paper	8	67	8	9

The results of Table 2 demonstrate that nonuniform particles present in a concentration as low as 0.1 weight percent provides enhanced mar and scratch protection (Sample 2); improvement is seen with spherical particles as well (Samples 6 and 7).

Table 3

Mar/Scratch Resistance Test	Sample 3 0.3% Zeeospheres 610	Sample 1 Control
BON AMI	73	61
1μ paper	88	55
9μ paper	15	8

The results of Table 3 demonstrate improved resistance with particles having an average particle size of 10μ (Sample 3).

Table 4

Mar/Scratch Resistance Test	Sample 4 0.3% WCA-3	Sample 5 0.3% WCA-3 + 1.0% Sunspheres 05	Sample 1 Control
BON AMI	88	98	57
1 μ paper	94	96	47
2 μ paper	90	89	50
9 μ paper	21	13	6

The results of Table 4 demonstrate that a blend of particles may also be used to improve mar and scratch resistance. The blend of particles (Sample 5) gave results, particularly for mar resistance, that were improved over those obtained with the individual particles, Sample 4 and Sample 6 (shown in Table 2). In all cases use of particles gave improved mar resistance when compared with control (Sample 1).

Example 2

Powder clearcoats were prepared as described for Sample 3 in Example 1 only using either 0.3 weight percent zinc oxide (Mohs hardness of 5, Aldrich Chemical, 0.5 μ average particle size), Sample 8, or 0.3 weight percent diamond particles (Mohs hardness of 10, GE Superabrasives, 6 μ average particle size), Sample 9. Sample 10 was prepared using the same components and weights as Sample 8, but with addition of the zinc oxide after extrusion. That is, the ZnO particles were mixed until incorporated into the formulation following the grinding step with a Henschel mixer. Test panels were prepared and tested as described above. Results are presented in Tables 5 and 6 below.

Table 5

Mar/Scratch Resistance Test	Sample 1	Sample 8 0.3% ZnO	Sample 10 0.3 wt.% ZnO (post added)
BON AMI	49	59	80
1 μ paper	61	64	65
2 μ paper	46	62	58
9 μ paper	6	6	5

Both the pre-added ZnO (Sample 8) and post-added ZnO (Sample 10) imparted greater overall mar and scratch resistance as compared to the control (Sample 1).

Table 6

Mar/Scratch Resistance Test	Sample 9 0.3 wt.% diamond	Sample 1
BON AMI	77	35
1 μ paper	88	53
2 μ paper	92	50
9 μ paper	84	6

The results of Table 6, when compared with the results shown for Sample 2 in Table 2, demonstrate that the higher the diamond particle weight percent the greater the resistance, as compared with control samples lacking any particles.

Example 3

Sample 11 was prepared as generally described in Example 1, using the amounts shown in Table 7. Test panels were prepared and tested, also as described above, and compared with Control Sample 1. These results are presented in Table 8.

Table 7

Description	Wt. %
GMA Functional Acrylic	63.03
DDDA	20.70
Acrylic Flow Additive	1.10
Benzoin	0.2
Wax C Micropowder	0.6
Tinuvin 144	2.0
CGL-1545	2.0
HCA-1	2.0
ARMEEN M2C	0.37
WCA3 alumina platelets	8.0

Table 8

Mar/Scratch Resistance Test	Sample 11 8 wt. % alumina	Sample 1
BON AMI	94	51
1 μ paper	98	66
2 μ paper	98	61
9 μ paper	60	8

- 5 The results of Table 8 demonstrate dramatic improvement in mar and scratch resistance when using 8 weight percent alumina platelets as compared with Control Sample 1, lacking the particles.

Example 4

- 10 Samples 12 and 13 were prepared as described above for Samples 8 and 10, only using an α -alumina, nonspherical particle having an average particle size of 0.5 μ instead of ZnO. Test panels were coated as described above. As demonstrated in Table 9, performance of pre-addition (Sample 12)

and post-addition (Sample 13) formulations greatly exceeded that of Control Sample 1.

Table 9

Mar/Scratch Resistance Test	Sample 12 0.3 wt.% alumina pre-added	Sample 13 0.3 wt.% alumina post-added	Sample 1
BON AMI	89	89	49
1 μ paper	74	74	61
2 μ paper	67	64	46
9 μ paper	7	7	6

5

Example 5

Samples 15-17 were prepared as described in Example 1 using the components and weight percents shown in Table 10.

10

Table 10

Description	Sample 15 Control	Sample 16 wt.% silica	Sample 17 wt.% silica
GMA acrylic resin ¹⁴	79.18	77.52	77.52
DDDA	17.41	17.02	17.02
Benzoin	0.38	0.37	0.37
Triphenyltinhydroxide catalyst	0.98	0.96	0.96
Wax C Micropowder	0.53	0.52	0.52
ModaFlow ¹⁵	0.90	0.88	0.88
Goresil 210 ¹⁶	--	2.21	--
Goresil 25 ¹⁷	--	--	2.21

¹⁴ Almatex PD 9060 produced by Anderson Development Company.

¹⁵ Modaflow, an acrylic copolymer flow additive anti-crater additive commercially available from Solutia, Inc.

¹⁶ Silica particles, average particle size 2 μ , largest particle size 10 μ , commercially available from C.E.D. Process Minerals, Inc.

¹⁷ Silica particles, average particle size 2 μ , largest particle size 5 μ , commercially available from C.E.D. Process Minerals, Inc.

Panels coated with Samples 15-17 were subjected to the BON AMI and steel wool tests described above. Results are presented in Table 11. Gloss retention, i.e. resistance, was greatly improved with both silicas; the smaller silica (Sample 17) gave a haze value that is more desirable in a clear coat application without compromising performance.

Table 11

Description	Sample 15	Sample 16	Sample 17
Initial 20° gloss	83	79	82
BYK Gardner haze	36	84	39
BON AMI (20 double rubs)	82	98	98
Steel wool 0000# grade double rubs (10x, light hammer)	57	96	91
Steel wool 0000# grade double rubs (5x, light hammer)	74	97	95

Example 6

- 10 Samples 18-21 were prepared and tested as those of Example 5, using the components and weight percent shown in Table 12.

Table 12

Description	Sample 18	Sample 19	Sample 20	Sample 21
GMA acrylic resin	79.76	77.98	--	--
GMA/IBoMA Acrylic Resin ¹⁸	--	--	81.63	83.49
DDDA	17.51	17.12	13.48	13.78
Benzoin	0.38	0.38	0.37	0.38
Triphenyltinhydroxide catalyst	0.91	0.89	0.89	0.91
Wax C Micropowder	0.53	0.52	0.52	0.53
Modaflow	0.91	0.89	0.89	0.91
Goresil 25	--	2.22	2.22	--

Samples 18 and 19 were formulated with GMA Acrylic Resin and Samples 20 and 21 with GMA/IBoMA Acrylic Resin, which has a lower RI than the GMA Acrylic Resin. Samples 19 and 20 contained Goresil 25, while Samples 18 and 21 provided a control lacking the particle. The results in Table 13 demonstrate that the Δ haze value can be reduced when using a resin that has a RI closer to that of the particle used. The Δ haze for the GMA resin (particle vs. no particle) was 16, while the Δ haze for the GMA/IBoMA was only 8. The samples containing particles according to the present invention had improved performance over control samples.

¹⁸ Acrylic copolymer with 40% glycidyl methacrylate and 60% isobornyl methacrylate.

Table 13

Description	Sample 18	Sample 19	Sample 20	Sample 21
Initial 20° gloss	83	81	79	80
BYK Gardner haze	24	40	33	25
BON AMI (20x)	50	80	44	33
BON AMI (40x)	64	85	72	53
Steel wool 0000# grade double rubs (5x, light hammer)	76	96	75	59

Example 7

- Samples 22-25 were prepared and tested as described in Example 5
5 using the components and weight percent shown in Table 14.

Table 14

Description	Sample 22	Sample 23	Sample 24	Sample 25
GMA acrylic resin	79.76	75.20	69.81	73.72
DDDA	17.51	16.51	15.33	16.19
Benzoin	0.38	0.36	0.33	0.35
Triphenyltinhydroxide catalyst	0.91	0.86	0.80	0.84
Wax C Micropowder	0.53	0.50	0.46	0.49
Modaflow	0.91	0.86	0.80	0.84
Goresil 25	--	5.72	5.31	--
Nanoparticles ¹⁹	--	--	7.17	7.57

¹⁹ Nanoparticles were obtained from Clariant Corporation and dispersed in an acid functional material.

Results presented in Table 15 demonstrate that the formulations comprising only micro-sized particles (Sample 23) and only nano-sized particles in acid functional siloxane (Sample 25) performed better than Control, and that the best overall performance was seen when both particles
5 were present (Sample 24).

Table 15

Description	Sample 22	Sample 23	Sample 24	Sample 25
Initial 20° gloss	82	77	77	82
9μ paper	22	33	56	50
3μ paper	38	53	75	77
2μ paper	77	87	93	81
BON AMI (20 times)	72	90	93	81
Steel wool 0000# grade double rubs (5x, heavy hammer)	77	100	96	82

Example 8

10 Samples 26-29 were prepared using an acid functional polyester resin containing the components in the weights shown in Table 16. The Samples were tested as described above, only using cold rolled steel panels with an iron phosphate pretreatment, obtained from ACT Laboratories.

Table 16

Description	Sample 26	Sample 27	Sample 28	Sample 29
Albester 5150 ²⁰	72.80	64.41	64.41	64.41
TGIC ²¹	5.30	4.69	4.69	4.69
SCX-819 ²²	3.04	2.69	2.69	2.69
PL-200 ²³	1.10	0.97	0.97	0.97
Benzoin	0.80	0.71	0.71	0.71
KC-59-9200 ²⁴	0.45	0.40	0.40	0.40
KH-97-3788 ²⁵	0.35	0.31	0.31	0.31
Monarch 1300 ²⁶	1.56	1.38	1.38	1.38
Vansil W-50 ²⁷	14.60	12.91	12.91	12.91
Goresil 25	--	11.53	--	--
Goresil 210	--	--	11.53	--
NABALOX 713-10 ²⁸	--	--	--	11.53

²⁰ Acid functional polyester resin, obtained from McWhorter Technologies.

²¹ Triglycidylisocyanurate, commercially available from Cytec Corporation.

²² Fungicidal acrylic, used as anti-crater additive, commercially available from Johnson Polymer.

²³ Flow additive, high molecular weight acrylic suspended in silica, commercially available from Eston Chemical, Inc.

²⁴ Actiron 32-057, commercially available from Synthron, Inc.

²⁵ Anti-crater additive, imide hydroxy urethane resin powder.

²⁶ Carbon black pigment, commercially available from Cabot Corporation.

²⁷ Extender, wollastonite, commercially available from R.T. Vanderbilt Company, Inc.

²⁸ α -alumina, average particle size 0.55 microns, commercially available from Baikowski International.

Table 17

Description	Sample 26	Sample 27	Sample 28	Sample 29
Initial 20° gloss	30	26	23	32
BYK Gardner haze	486	475	483	479
9μ paper	32	44	54	46
3μ paper	64	83	88	84
2μ paper	104	126	126	117
BON AMI (10 x)	71	106	99	88
Steel wool 0000# grade double rubs (30x, heavy hammer)	94	126	135	109

As demonstrated in Table 17, the formulations of the present invention (Samples 27-29) performed better in all tests as compared with the Control,

- 5 Sample 26. A variety of resin types, including those containing pigments, are suitable for use in the present invention.

Example 9

- 10 Liquid coating compositions (Samples 30-32) were prepared using the components listed in Table 18.

Table 18

Description	Sample 30	Sample 31	Sample 32
Nanoparticles ²⁹	4.50	4.41	4.31
Methyl amyl ketone ³⁰	24.30	23.78	23.29
Acrylic resin ³¹	37.75	36.95	36.18
Solvent ³²	5.74	5.62	5.50
Butyl cellosolve acetate ³³	1.04	1.02	1.00
Particle paste ³⁴	--	2.13	4.17
Isocyanate crosslinker ³⁵	26.60	26.04	25.49
Tin catalyst ³⁶	0.06	0.06	0.06

The components of the particle paste were sealed in an eight ounce jar and shaken on a paint shaker for 3.5 hours to disperse the particle paste.

- 5 The grind media was filtered out and the material was ready to use. The above ingredients were mixed and sprayed within 10 minutes due to the short pot life, which is normal for refinish two-pack systems.

The ED5051 black primer panels described in Example 1 were hand sprayed at 45 psi, 71.6°F, and 63 percent relative humidity, and air cured.

- 10 Panels were tested after one week to provide sufficient cure. The tests performed and results are shown in Table 19.

²⁹ 30% nano silica particles/70% siloxane.

³⁰ Eastman Chemicals.

³¹ Acrylic resin having hydroxyl functionality.

³² Exxate 600 solvent (hexyl acetate) from Union Carbide.

³³ Union Carbide.

³⁴ 63.4% particle paste prepared by mixing the following:

Goresil 25	56.0 grams (35.65%)
Methyl amyl ketone	21.0 grams (13.37%)
Acrylic resin having hydroxyl functionality	10.0 grams (3.37%)
Solsperse 2400, commercially available from Avecia	0.10 grams (0.06%)
1 mm Zircoa beads	70.0 grams (44.56%)

³⁵ HDTLV, from Rhodia Inc.

³⁶ Metacure T-12 catalyst, commercially available from Air Products and Chemicals, Inc.

Table 19

Description	Sample 30	Sample 31	Sample 32
Initial 20° gloss	83.1	81.8	81.7
9μ paper	51	46	67
3μ paper	66	73	78
2μ paper	84	90	89
BON AMI	79	96	94
Steel wool 0000# (10 x); Atlas tester	62	84	76

Use of the present particles in the liquid coating system (Samples 31 and 32) imparted improved mar and scratch when compared with the Control (Sample 30).

Example 10

Panels were coated and tested as described in Example 1, using the coatings set forth in Table 20. Particle load was 0.3% for Samples 34-36, and 0.1% for Sample 37. The panels were subjected to QUV exposure for 500, 1000 or 1500 hours according to ASTM D-4587. As illustrated in the table, the present compositions containing particles showed improved scratch resistance following QUV exposure as compared with the control lacking particles, and in many cases scratch resistance improved as the length of QUV exposure increased. The results in the table are given in % gloss retention using 20° gloss.

103027 01120001

Table 20

	Sample 33 DJ55 ³⁷	Sample 34 DJ55 W210	Sample 35 DJ55 WCA3	Sample 36 DJ55 Goresil 25	Sample 37 DJ55 Diamond Dust MBM 4-8
2μ paper initial	62.6	72.2	90.3	72.9	85.9
500 hours	77.4	94.1	96.9	95.6	93.1
1000 hours	76	94.7	94.3	91.7	94.6
1500 hours	92.8	92.2	97.4	97.6	91.1
9μ paper initial	13.6	20.1	28.2	19.0	77.7
500 hours	11.7	17.6	45	17	77.7
1000 hours	21.9	20	57	30.4	82.4
1500 hours	40.1	35.6	72.5	42.7	68.1

³⁷ Acrylic powder coating, commercially available from PPG Industries, Inc.

Samples 38-44 were prepared and tested as described in Example 5, using the components shown in Table 21. The use of heat-treated particles (Samples 42, 43 and 44) generally gave better results than their non-heat treated counterparts (Samples 39, 40 and 41, respectively), which still gave better performance overall than control Sample 38, which had no particles.

5

Table 21

	Sample 38	Sample 39	Sample 40	Sample 41	Sample 42	Sample 43	Sample 44
GMA acrylic resin	79.76	77.98	77.98	77.98	77.98	77.98	77.98
DDDA	17.51	17.12	17.12	17.12	17.12	17.12	17.12
Wax C Micropowder	0.53	0.52	0.52	0.52	0.52	0.52	0.52
Benzoin	0.38	0.37	0.37	0.37	0.37	0.37	0.37
Triphenyltrihydroxide catalyst	0.91	0.89	0.89	0.89	0.89	0.89	0.89
Modaflow	0.91	0.89	0.89	0.89	0.89	0.89	0.89
Sunspheres 05	--	2.22	--	--	--	--	--
T64-20 ³⁸	--	--	2.22	--	--	--	--
Zeospheres W210	--	--	--	2.22	--	--	--
HT ³⁹ Sunspheres 05	--	--	--	--	2.22	--	--
HT T64-20	--	--	--	--	--	2.22	--
HT Zeospheres W210	--	--	--	--	--	--	2.22

³⁸ Tabular alumina, 7μ average particle size, maximum particle size 20μ, commercially available from Alcoa.

³⁹ "HT" = heat treated for three hours at 700°C.

Table 21 (Cont'd)

	Sample 38	Sample 39	Sample 40	Sample 41	Sample 42	Sample 43	Sample 44
Initial 20° gloss	83.0	82.8	79.1	77.1	82.7	79.1	76.9
9μ paper	13.6	16.7	62.5	44.4	28.4	67.8	50.3
3μ paper	31.1	29.7	83.3	77.4	57.9	88.5	76.7
BON AMI (20x)	79.4	93.8	92.6	97.5	96.4	94.6	98.1
Steel wool 0000# gauge (5x, light hammer)	83.1	87.8	92.8	98.8	95.8	97.7	98.8
Steel wool 0000# gauge (5x, heavy hammer)	83.0	87.6	91.2	97.1	90.7	95.2	97.5

Whereas particular embodiments of this invention have been described above for purposes of illustration, it will be evident to those skilled in the art the numerous variations of the details of the present invention may be made without departing from the invention as defined in the appended claims.